

This article was downloaded by: [Tomsk State University of Control Systems and Radio]

On: 20 February 2013, At: 12:36

Publisher: Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954

Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



## Molecular Crystals and Liquid Crystals

Publication details, including instructions for authors and subscription information:

<http://www.tandfonline.com/loi/gmcl16>

### Overhauser Shift on Radical Cation Salts of Fluoranthene

G. Denninger<sup>a</sup>, W. Stöcklein<sup>a</sup>, E. Dormann<sup>a</sup> & M. Schwoerer<sup>a</sup>

<sup>a</sup> Physikalisches Institut und Bayreuther Institut für Makro-molekülforschung (BIMF), Universität Bayreuth, D-8580, Bayreuth, BRD

Version of record first published: 17 Oct 2011.

To cite this article: G. Denninger, W. Stöcklein, E. Dormann & M. Schwoerer (1985): Overhauser Shift on Radical Cation Salts of Fluoranthene, *Molecular Crystals and Liquid Crystals*, 120:1, 233-236

To link to this article: <http://dx.doi.org/10.1080/00268948508075792>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.tandfonline.com/page/terms-and-conditions>

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages

whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

## OVERHAUSER SHIFT ON RADICAL CATION SALTS OF FLUORANTHENE

G. DENNINGER, W. STÖCKLEIN, E. DORMANN and M. SCHWOERER  
Physikalisches Institut and Bayreuther Institut für Makro-  
molekülforschung (BIMF), Universität Bayreuth, D-8580 Bay-  
reuth, BRD

Abstract Hyperfine interaction, proton spin relaxation and  
dynamic proton spin polarization were determined for the  
radical cation salts of fluoranthene:  $(\text{FA})_2^+\text{PF}_6^-$  and  $(\text{FA})_2^+\text{SbF}_6^-$   
by observation of the Overhauser shift.

### INTRODUCTION

The radical cation salts  $(\text{FA})_2^+\text{PF}_6^-$  and  $(\text{FA})_2^+\text{SbF}_6^-$ , respectively are  
conducting crystals with a good 2:1 stoichiometry. They show a  
quasi metallic phase above  $T_c \sim 180$  K and a semiconducting phase<sup>1</sup>  
below  $T_c$ . Their strong single line ESR spectrum originates from  
the charge carriers<sup>2,3</sup>. It is motionally narrowed with a width  
 $\Delta H_{\text{pp}} = 10 \text{ mGauss}$  for  $T = 300$  K and  $7.5 \text{ mGauss}$  for  $T = 190$  K. Due to  
this extremely narrow line its shift due to the nuclear field of  
the protons (Overhauser shift) can be analyzed in detail<sup>4</sup>.

### EXPERIMENT

After recording a conventional ESR spectrum the external field  $B_0$  is  
fixed to the center of the line. By applying an additional radio-  
frequency field and sweeping through the range of the proton Lar-  
mor frequency a shift of the ESR resonance field is observed. This  
shift  $\Delta B_{\text{ov}}$ , predicted by Overhauser<sup>5</sup>, is due to switching off the

average nuclear field of the protons by saturating the proton NMR. In first order approximation the shift  $\Delta B_{ov}$  produced by  $N$  protons ( $N = 20$  for  $(FA)_2PF_6$ ) per electron is given by

$$\Delta B_{ov} = -(2 \cdot g_e \mu_B)^{-1} \cdot N \cdot \mathcal{P} \cdot \bar{A}_{zz} \quad (1)$$

where  $\bar{A}_{zz}$  is the averaged  $z$ -component of the proton's hyperfine tensors. If the ESR is saturated the polarization  $\mathcal{P}$  is a dynamic nuclear polarization (DNP) and thus exceeds the thermal nuclear spin polarization  $\mathcal{P}_0$ .

## RESULTS

1. The magnitude of the shift ( $\approx 1$  mGauss at  $T = 293$  K for  $(FA)_2PF_6$ ) yields the product  $\mathcal{P} \cdot \bar{A}_{zz}$ . We observe a dependence on the incident microwave power  $P$ :

$$\Delta B_{ov} \propto \mathcal{P}_0 \cdot [1 + V \cdot \frac{\alpha P}{1 + \alpha P}] \quad (2)$$

$V = \mathcal{P}_{max}/\mathcal{P}_0 - 1$  is the Overhauser enhancement factor. By measuring  $\Delta B_{ov}(P)$  down to small microwave powers  $P$  the enhancement factor  $V$  was determined. For  $(FA)_2PF_6$  at  $T = 185$  K we deduced a value of  $V = 400 \pm 60$ . This yields  $\bar{A}_{zz}/g_e \mu_B = -1.1$  Gauss.

2. After sudden switch-off of the rf-field a monoexponential decrease of the shift is observed (inset

fig.1). It can be shown experimentally and theoretically that one observes directly the longitudinal spin lattice relaxation time  $T_1^P$  of the protons. Fig. 1 shows  $1/T_1^P$  versus  $T$ . The straight line indicates the Korringa law  $T_1^P \cdot T = \text{const.}$  to be obeyed for  $T > 200$  K. Below  $T_c$  the relaxation rate rises because further relaxation me-

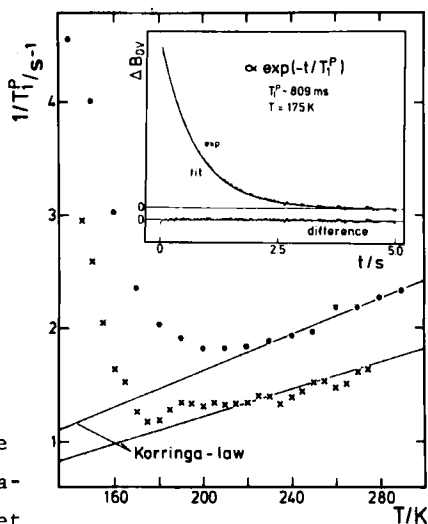


FIGURE 1 Proton  $1/T_1^P$  versus  $T$  determined by the Overhauser shift method. (●)  $(FA)_2SbF_6$ , (×)  $(FA)_2PF_6$

chanisms for the protons<sup>6</sup> become important. A consequence is the decrease of  $\Delta B_{ov}$  (fig.3).

3. By applying rf-pulses of variable length  $\tau$  the DNP can be manipulated as usual in NMR spectroscopy. Fig. 2 shows the variation of the ESR line shift with the pulse length  $\tau$ . The experimental data (•) are fitted by:

$$\Delta B_{ov}(\tau) = \Delta B_{ov}(\infty) [1 - \cos\{\frac{\pi \cdot \tau}{T_{180}^0}\} \cdot \exp(-\tau/2 \cdot T_2^{eff})] \quad (3)$$

where  $\Delta B_{ov}(\infty)$  is the shift for an infinitely long rf pulse.  $T_{180}^0$  is the time for an  $180^\circ$ -pulse. The fit yields  $T_2^{eff} \approx 16 \mu s$ . The value is in good agreement with the  $T_2^{eff}$  deduced from the line width of the shift signal ( $T_2^{eff} \approx 13 \mu s$ ).

4. It should be pointed out that all the results are obtained with tiny single crystals and show a clear anisotropy with respect to the crystal axis. We observe  $T_{1\perp}^p/T_{1\parallel}^p = 1.15$  and  $(\Delta B_{ov\perp})/(\Delta B_{ov\parallel}) \approx 3$ , where " $\perp$ " and " $\parallel$ " denote the direction of the needle axis with respect to  $B_0$ .

5. The temperature dependence of  $\Delta B_{ov}$  shows a strong variation near  $T \approx 185$  K. Above 190 K one observes  $(1/\Delta B_{ov}) \propto T + \text{const.}$  Below 190 K the shift decreases similar to the decrease of the ESR-intensity (fig. 3).

6. The ratio  $\nu_e/\nu_p$  is independent on the magnetic field which acts on the electrons and protons. Thus the anisotropy of  $\nu_e/\nu_p$  reflects the anisotropy of the spin orbit coupling and is unaffected by demagnetization effects.

7. No shift  $\Delta B_{ov}$  induced by the phosphorus or fluorine nuclei was

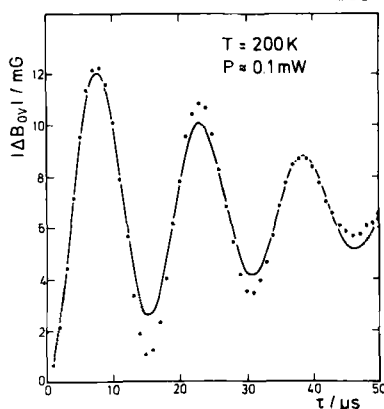


FIGURE 2 shift  $\Delta B_{ov}$  versus rf-pulse length  $\tau$

detected outside the error limit of 0.5 % of the proton value.

### CONCLUSION

The Overhauser shift method is highly sensitive to determine the proton spin relaxation times  $T_1$  and  $T_2^{\text{eff}}$  and their angular dependence in tiny single crystals of organic conductors.

Furthermore the Overhauser enhancement and the average hyperfine tensor components can be determined directly. Thus the experiments yield specific informations

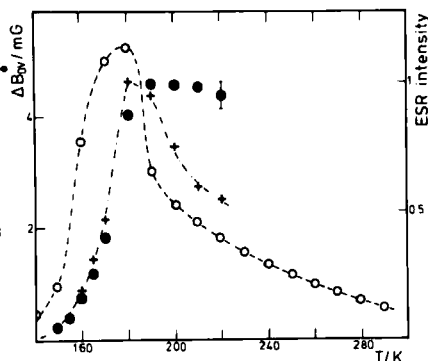


FIGURE 3  $\Delta B_{\text{Ov}}$  versus  $T$  for  $(\text{FA})_2\text{PF}_6$  ( $\bullet$ ) and  $(\text{FA})_2\text{SbF}_6$  ( $\circ$ ) ESR intensity for  $(\text{FA})_2\text{PF}_6$  (---)

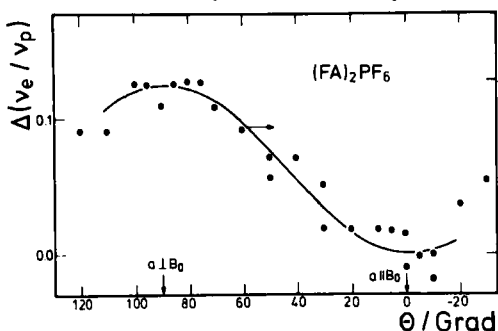


FIGURE 4 ( $\bullet$ ):  $\Delta(v_e/v_p)(\theta) = v_e/v_p(\theta) - v_e/v_p(0)$   
solid line:  $\Delta g(\theta) = g(\theta) - g(0)$

on the interactions of the almost free charge carriers.

### ACKNOWLEDGEMENTS

We acknowledge valuable discussions with M. Mehring, J.U. v. Schütz and K.P. Dinse. This work was supported by Stiftung Volkswagenwerk

### REFERENCES

1. Ch. Kröhnke, V. Enkelmann, G. Wegner, *Angew. Chemie* 92, 941 (1980).
2. H. Eichele, M. Schwoerer, Ch. Kröhnke, G. Wegner, *Chem. Phys. Lett.* 77, 311 (1981).
3. G. Sachs, W. Stöcklein, B. Bail, E. Dormann, M. Schwoerer, *Chem. Phys. Lett.* 89, 179 (1982).
4. G. Denninger, W. Stöcklein, E. Dormann, M. Schwoerer, *Chem. Phys. Lett.* 107, 222 (1984).
5. A.W. Overhauser, *Phys. Rev.*, 92, 411 (1953).
6. W. Höptner, M. Mehring, J.U. v. Schütz, H.C. Wolf, B.S. Morra, V. Enkelmann and G. Wegner, *Chem. Phys.* 73 (1982) 253.